Lecture 8

hor mones

Steroidal Hormones

Biosynthesis of steroids:



N.B.

- CYP 17 is described as 17-a-hydroxylase C20-21 lyase. It's like 2 enzymes in one g First it introduces a hydroxyl to carbon 17 (hydroxylase), then it removes carbon 20 & 21 (lyase) leading finally to formation of a ketone.
- The major Progestagen g Progesterone.
- The major gonadal Estrogen \rightarrow Estrone Estradiol.
- The major gonadal Androgen \rightarrow Testosterone.
- The major Mineralocorticoid [control electrolytes] \rightarrow Aldosterone.
- The major Glucocorticoid [control metabolism] \rightarrow Corticosterone Cortisol \rightarrow varied between Species.

General Chemistry of Steroidal Hormones



Conformational aspects of steroids:18

Substituents are:

- Axial (a) or equatorial (e). •
- Above (α) or below (β) the plane of the ring.

The ring fusion in natural hormones is:

- Trans/trans/trans.
- Cis/trans/trans in few examples.
- Trans A/B rings trans fusion may be referred to as $5-\alpha$.
- Trans or cis is determined by the substituents on the bridging carbons. eq: In pregnane. For trans bridging between A & B, susbstituents on carbon 10 (i.e methyl) and carbon 5 (i.e H) should be trans to each other. The methyl at carbon 10 is Beta thus the hydrogen on carbon 5 should be alpha in order to be trans (that's why they call it 5 alpha)

Default g The substituents on carbons 10,13,17 are b and the rings are in trans

Nomenclature:

Nomenclature rules:

- 1. <u>Trans A/B rings fusion</u> is understood from writing the name of the carbocycle. A change in A/B rings fusion is indicated before the name of the carbocycle
- 2. <u>Angular methyls</u> (18 & 19) & the side chain at 17 are usually \rightarrow <u>beta</u> oriented.
- 3. The carbocycle has to be drawn with ring <u>A bottom left</u>, and <u>ring D top right</u>.
- 4. The location and orientation of substituents should be indicated.
- 5. Single substitutent is written as suffix.
- 6. <u>2 substituents</u>: one prefix, one suffix according to priority role.
- 7. More than one substituent \rightarrow alkyl-halo-nitro as prefixes \rightarrow others suffixes in alphabet order.

Entamproo		
Estradiol	Testosterone	Cortisone
HO HO	CH ₃ H CH ₃ H H H H	
Estra-1,3,5(10)-triene- 3,17-β-diol	17β-Hydroxyandrost-4- ene-3-one	17,21-Dihydroxypregn-4-ene- 3,11,20-trione

Examples:

<u>Sex (Gonads) Hormone</u> Female Sex hormones(Estrogens & Progestins)

Long acting estradiol esters			
Most are oil solub	Most are oil soluble given I.M. [prodrugs] g They are long acting because they are		
prodrug that slow	ly release the active form and they are use	ed during the i	menopause
		R	R'
	1. <u>Estradiol benzoate</u> (Menovis) g 1 week duration.	Н	O II —C—Ph
	2. Estradiol dipropionate	О Ш —С—СН ₂ —СН ₃	О Ш —С—СН ₂ —СН ₃
H ₃ C V	 <u>Estradiol cypionate g</u> cyclopentyl propanoate ester 	0 C−CH ₂ −∕	Н
R'O	 4. <u>Estradiol</u> <u>hexahydrobenzoate</u> (Estra-1,3,5(10) triene-3,17β-diol-17- pentanoate) 		Н
	5. <u>Estradiol Valerate</u> (Progynova) orally bioavailable, Used as <u>estradiol</u> <u>replacement therapy.</u>	0 II —C—(CH ₂) ₃ —CH ₃	Н

Lecture [8]

Conjugated (attenuated) estrogens			
Weakened g Better than using potent ones to avoid the breast cancer and i the side effects.			
Estrone Na sulfate	<u>Equilin Na sulfate</u> <u>(Premarin)</u>	<u>Estropipate</u>	
H ₃ C H ₃ C	NaO ₃ SO		
NaO ₃ SO	Na 17-Oxo-estra1,3,5(10), 7-tetraene-3- sulfate	Piperazine 17-oxo-estra-1,3,5(10)- triene-3-yl-sulfate	
Water soluble \rightarrow orally bioavailable		More stable with acids	
<u>Used as</u> \rightarrow <u>Hormonal replacement</u> to treat postmenopausal climacteric symptoms &			
Deservative a service of both (estrong of service No. CO.) to be a settion.			
<u>Premarin</u> [®] g consist of both (estrone & equilin Na ₂ SO ₄) taken together. They ppt by the			
HCL in the stomach but still they are taken because not all will be ppt, some will not.			
Estropipate: It is a piperazine sulfate. It contains 2 basic nitrogens and needs more acid			
to neutralize thus more stable			

Synthesis of estradiol and estrone derivatives***:

Lecture [8]

<u>Non-steroidal estrogens</u>			
[estrone nucleus is NOT necessary for activity]			
Diethylstilbsterol (DES)	Fosfosterol	Dienesterol	
H ₃ C H ₃ C H ₁ C	H ₃ C H ₃ C CH ₃ NaO ₃ PO	H ₃ C H ₃ C H ₃ C H ₀ Cis	
trans isomer is 10X as estrogenic as its cis isomer [see below***]	<u>more selective</u> used for <u>menopausal</u> <u>symptoms & prostate cancer</u>	used as <u>vaginal cream</u> for postmenopausal vaginal atrophy	

Estradiol	Trans-Diethylstilbsterol	Cis-Diethylstilbsterol
11.9 A H ₃ C HO	12.9 A HO HO HO H ₃ C	HO HO HO H ₃ C

Estrogen antagonists [Selective Estrogen Receptor Modulators] (SERM)

A. <u>Triphenylethylene derivatives</u> We can't describe it as anti estrogenic in a general term because of the geometrical isomers Z, E differ in the activity as Z is the agonist and E is the antagonist in a particular tissues so term SERM is used instead

lein			
Clomiphene	Tamoxiphene	Toremiphene	
(Clomid ^â)	(Nolvadex ^â)	(Fareston ^â)	
<u>The dimethyl ethoxy amine is</u> responsible for antagonist action. CI <u>CI</u> <u>CI</u> <u>CI</u> <u>CI</u> <u>CI</u> <u>CI</u> <u>CI</u>	N-desmethyl metabolite contribute to the antiestrogenic activity	CI	
• Mix of <u>Z (cis)</u> \rightarrow estrogenic	• <u>Z (trans) isomer</u>	\rightarrow Estrogenic in uterus \rightarrow	
& <u>L (trans)</u> → anti-estrogenic.			
Used in uterus - hypothalamus -	Used in treatment o	or <u>bresat cancer</u> .	
Anovulatory infertility	 T Risk of uterine ca 	rcinoma.	

<u>Clomiphen</u> g [Anovulatory means decrease in ovulation]. E isomer is used to enhance the ovulation but the side effect is poly pregnancy. It is used to enhance the ovulation by sending false signals to the hypothalamus g h GnRH g h in the LH and the FSH g h estrogen

Synthesis of Clomiphene:

Lecture [8]

B. Aromatase (CYP 19) inhibitors		
Anstrozole (Arimidex ^â)	Letrozole (Femara ^â)	
N H ₃ C CH ₃ H ₃ C CH ₃ H ₃ C		
By the Coordination of the CYP 19 heme iron g by the chelation (due to triazole) of the		
heme of the enzyme and this will prevent the aromatization.		
Used for the treatment of breast carcinoma		

Aminoglutethimide (Cytadren ^â)	Formestane (Lentaron ^â)	Testolactone (Teslac ^â)
	CH+ OH <u>4-hydroxy androstanedione</u>	CH ₃ C O O
 <u>Non-specific</u> aromatase inhibitor. Used in ttt of <u>Cushing`s syndrome</u>. 	 With <u>androstane</u> skeleton. I.M. injection every 2 weeks. Used in the treatment of the <u>breast cancer</u>. 	 <u>Androstane</u> skeleton. <u>Non-specific</u> inhibitor of CYP19. <u>Orally</u> 4 times daily in large doses. It's <u>anabolic</u> steroid.

Synthesis of Letrozole:

hor mones

Lecture 9

Steroidal Hormones

Progestins, Progestogens or Progestional agents

Metabolism of progesterone:

Progesterone is:

- Biosynthesized in ovaries testis adrenals → mainly under the control of LH [Progesterone has a <u>negative feed back on LH</u>].
- <u>Orally inactive</u> due to <u>1st pass & bacterial decomposition</u> of ring D & reduction of the –CO- of ring A and ring D.
- A key intermediate in the biosynthesis of corticosteroids-androgens then estranes.
- Acts (complementarily with estrogens) to control menstrual bleeding-release of ovum - preparing the uterus to receive fertilized ovum_suppression of ovulation during pregnancy-maintenance of pregnancy through depressing uterine contraction.

Assay of progesterone:

• By INH \rightarrow give a color product g colorimetry.

Uses of progestins:

- 1. Functional uterine bleeding habitual and threatened abortion.
- 2. Endometriosis.
- 3. Anti-androgen for \rightarrow treatment of prostate cancer.
- 4. Adjuvant therapy in endomtrial carcinoma.
- 5. Contraceptive \rightarrow as single treatment or in combination with estrogens.

<u>17-a-Hydroxyprogesterone derivatives</u>

- 17α -hydroxy function to protect the -CO- from 17 keto reductase \rightarrow enhance metabolic stability & oral bioavilability [Protection of a-phase]
- Make use of the –OH to form <u>oil soluble esters</u> (produgs, long acting).
- <u>Blocking position 6 of ring B and/or introduction of unsaturation in ring B</u> will enhance metabolic stability, too.

Hydroxyprogesterone caproate (Primolut-depot ^â)	Medroxyprogesterone acetate (Depo-Provera ^â)	Megesterol actate (Megace ^â)
H ₃ C CH ₃ , 1110 C ₅ H ₁₁	CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃	CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃
3,20-Dioxopregn-4-ene- 17α-yl caproate	17-Hydroxy-6α-methylpregn-4- ene-3,20-dione acetate	17-Hydroxy-6-methylpregn- 4,6-diene-3,20-dione acetate
 Once weekly injection. Used in <u>habitual or threatened</u> abortion. 	 Injection every 3 months. <u>Used in:</u> Progestin only contraceptive. Management of endometrial carcinoma. Recently introduced monthly co-injection with estradiol cypionate (Lunelle). 	Used as <u>tablets</u> in treatment of <u>breast &</u> <u>endometrial carcinoma</u> .

Medroxyprogesterone acetate: acetate ester will lead to:

• The slow release of the17 hydroxy progesterone.

• Resistant to the metabolism because of protecting function on the $17-\alpha$ position.

So these TWO factors will lead to prolonging the action up to three month

Megesterol actate:

It has been found that the progestins can compete with estradiol for the estrogenic receptors (antagonist action) so this compound which has the following properties:

- Hydroxy ester at C₁₇.
- Double bond at C₆.
- Methyl at C₆ to block metabolism

g can be used for the oral treatment of the breast cancer & endometrial carcinoma. [breast cancer & endometrial carcinoma is due to increased estrogen].

Lecture [9]

Cyproterone acteate (Androcur ^â)	Dydrogestone (Duphaston ^â)
CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ CH ₃	CH ₃ O
6 -Chloro-17-hydroxy-1- α ,2- α - methylenepregn-4,6-diene-3,20- dione acetate	$9-\beta$,10 α -Pregn-4,6-diene-3,20-dione
<u>Uses:</u>	Used as tablets in ttt of <u>irregular duration &</u> <u>occurrence of period</u> .
Anti-androgenic g treatment of prostate cancer.	<u>Changing conformation of progesterone</u> \rightarrow orally active-no contraception-No virilizing action on fetus

Cyproterone acteate:

<u>Methylene bridge</u> will yield affinity to the androgenic receptors (compete with testosterone & hydroxycortisone) g so it is used in the treatment of the <u>prostate cance</u>

17a-Ethinyl-19-Nortestosterone derivatives

- They are orally active.
- They are progestogenic agents by activity but not chemically
- Angular C₁₉ is not essential for progestional activity.
- Introduction of <u>17a- ethynyl</u> function OR <u>17a- unsaturation</u> give ↑ oral activity- ↑ metabolic stability - ↓ androgenic activity.
- 17α-OH allows the formation of <u>long acting esters</u>.

• Norelgestromin used as contraceptive cutaneous patches with estradiol.

- All are used as <u>oral contraceptive</u> in combination with estrogens.
- BUT <u>levonorgestrel is progesterone only type contraceptive</u> & may be used as implants or to impregnate IUD.

Anti-Progesterone

Androgens and Anabolics

Metabolism of testosterone

Testosterone:

- <u>Short duration</u> of action.
- Orally Inactive (not bioavailable) → due to attack & decomposition of ring D by the liver & intestinal bacterial.
- Have anabolic/androgenic activity.

Testosterone esters		
		R
о Д	j Testosterone propionate [short onset & duration]	$-C_2H_5$
CH ₃ H	k Testosterone phenyl propionate [intermediate onset & duration]	$\neg - \bigcirc$
	<u>1 Testosterone isocaproate</u> [Long onset & duration]	CH ₃ CH ₃
5	m Testosterone heptanoate (enanthate) [long acting]	CH3

- <u>- onset & duration</u> according to the size of the ester R.
- Used as a single agent OR mixture.
- It's <u>Prodrugs</u> \rightarrow Given by <u>I.M. injections</u>.

 <u>Sostanone ampoule</u>[®] is composed of 1st three esters g given in a <u>single injection</u> g short onset of the propionate + long duration of action of isocaproate.

Uses of Androgens & anabolics:

- 1. Delayed puberty & growth in males.
- 2. Impotence & male climacteric.
- 3. Osteoporosis in males & females.
- 4. Debilitating illness and anemia.

Orally bioavailable testosterone derivatives (anabolic/androgenic agents)

<u>By:</u>

- 1. Introduction of <u>17a-methyl</u> $\rightarrow \uparrow$ metabolic stability and oral activity.
- 2. <u>- length of the alkyl [>Me]</u> $\rightarrow \downarrow$ both anabolic & and rogenic activity.
- 3. Introduction of <u>17a–ethynyl</u> \rightarrow imparts <u>progestogenic</u> activity (ethisterone).
- 4. <u>Alkylation at positions 1,2, 7 or 18</u> \rightarrow \uparrow anabolic/androgenic ratio.
- 5. Introduction of <u>4b &/or 11b-hydroxyl</u> in methyltestosterone introduction of <u>F at C₉</u> or <u>unsaturation to ring A</u> \rightarrow <u>- the anabolic & ⁻ androgenic activity</u>.

Synthesis of methyltestosterone:

<u>N.B.</u>

- Oppenour oxidation → oxidation of a 2ry alcohol with a ketone (cyclohexanone)
 + Al isopropoxide → formation of -4-en-3-one conjugated system.
- There is <u>no pure</u> androgenic nor pure anabolic action.

Mainly androgenic		
Ovymastrona (aranahalâ)	Methandrostenolone	Fluoxymestrone
Oxymestrone (oranabor)	(Dianabol ^â)	(Ultandren ^â)
	CH ₃ H CH ₃ H H H H H H H H H H H H H H H H H H H	$HO = H + CH_3 OH + H_3C + H + H_3C + H + H_3C + H + H + H + H + H + H + H + H + H + $
4, 17β-Dihydroxy-17- methylandrost-4-en-3-one	17β-Hydroxy-17- methylandrost-I,4-dien-3-one	9-Fluoro-11β,17β-dihydroxyl- 7-methylandrost-4-en-3-one 20X as anabolic – 10X as androgenic as methyl testosterone
Methenolone (primoblan ^â)	Oxymetholone (Anabolon ^â)	Stanozolol (Stromba)
CH ₃ CH ₃ C		CH ₃ OH CH ₃ H HN HN Azasteroid
1-Methyl-3-oxo-androst -1-en-17β-yl acetate Methonolone acetate (Primoblan depot ^â)	17β-Hydroxy-2- hydroxymethylene-17-methyl androstan-3-one	17α-Methyl-2`H-androst- 2-eno[3,2-c]-pyrazol-17-ol

Anti-androgens and 5a-reductase inhibitors

Uses:

- 1. <u>Males</u> \rightarrow hypersexuality benign prostatic hyperplasia prostaic carcinoma.
- 2. <u>Females</u> \rightarrow treatment of acne hirsutism alopecia androgenetica.

Chemical nature:

- <u>Steroids</u> \rightarrow e.g. cyproterone acetate spironolactone.
- <u>Non-steroidal</u> \rightarrow e.g. cimetidien flutamide and its derivatives.

Flutamide (Flutamex ^â)	Nilutamide (Nilandron ^â)
$\xrightarrow{H_{3}C} 0 \xrightarrow{F} F_{F}$ $\xrightarrow{H_{3}C} HN \xrightarrow{I} NO_{2}$ <u>Metabolized to hydroxyl</u> <u>flutamide [active]</u>	$HN \rightarrow V$ $H_3C \rightarrow V$
3'-trifluoromethyl-4`-nitro- isobutyranilide	1-(3'-Trifluoromethyl-4'-nitrophenyl) -4,4- dimethylhydantoin
$t_{1/2} = 8 hrs$	$t_{1/2} = 50 \text{ hrs}$

Lecture 10

hor mones

Steroidal Hormones

Mineralocorticoids and Glucocorticoids

[i] Mineralocorticoids

- Biosynthesis is stimulated by ACTH, Angiotensin II and serum K+.
- Enhances Na⁺, H₂O re-absorption, K⁺ excretion.

Fludrocortisone acetate

<u>9-Fluoro-11β,17,21-trihydroxypregn-4-ene-3,20-dione</u>

- <u>9 a- Fluoro substituent</u> h mineralocorticoid activity 500X and glucocorticoid activity 8 X as hydrocortisone.
- <u>Mineralocorticoid</u> used to treat <u>adrenal insufficiency</u> e.g. <u>Addison's disease</u> and to treat <u>orthostatic hypotension</u>.

[ii] Glucocorticoids

Uses of Glucocorticoids:

- <u>Immunosuppressant</u> in case of inflammatory diseases e.g. psoriasis, anaphylactic shocks, severe asthma, poison ivy dermatitis, ulcerative colitis, and rheumatoid Arthritis.
- Anti-allergic.
- Leukemia, lymphoma.
- Nausea associated with chemotherapy.

Cautious:

Diabetes, hypertension, water retention, infectious diseases and sudden withdrawal particularly after long term therapy, glaucoma, osteoporosis.

SAR of glucorticoids:

- They are all pregnane derivatives with the all trans ring fusion.
- Introduction of <u>unsaturation at $C_1 g h$ glucocorticoid activity</u>.
- For <u>all glucocorticoids and mineralocorticoids</u> an <u>11b-hydroxyl function is</u> <u>essential for activity</u>.
- <u>The 9a- fluoro substituent</u> will h ionization of 11β- hydroxyl function by inductive action, this enhances interaction with the receptor through H bonding. However, the 9-fluoro substituent will h glucocorticoid activity.
- <u>16a-hydroxyl or methyl</u> will negate the sodium retention of the 9α fluorine.
- <u>21 halo substituent</u> greatly enhances lipophilicity and glucocorticoid activity, due to enhancement of the receptor binding, delays rate of dissolution and prolongs the duration of action.

- <u>All are ∆1 corticosteroids</u>: 4-5 X as glucocorticoid and nearly equal mineralocorticoid to hydrocortisone.
- <u>Ring A conformation</u>: [Saturated g Chair] [4-en-3-one g half chair] [1,4-dien-3-one g flattened boat]

Triamcinolone:

With the <u>highest hydrophilicity</u>, <u>lowest oral bioavailability</u> and glucocorticoid activity but all are of <u>better glucocorticoid and lower mineralocorticoid activity</u> relative to hydrocortisone.

Steroidal Prodrugs to improve water solubility and shorten onset of action

- Injection in acute pain, asthma and life saving conditions.
- Onset = 2 minutes in vivo, but 2 years on shelf!!!

Steroidal Prodrugs to improve lipid solubility, prolong the duration of action and/or topical applications

Topically only applied steroids

- Dihalo, halo-ester, diester corticosteroids Lipophilic (log p >4.5).
- With low H₂O solubility, Minimal absorption.

Beclomethazone dipropionate	Fluocinolone acetonide	Fluticasone propionate
OCO-C ₂ H ₅ HO CH ₃ H CH ₃ H CH ₃ H CH ₃ CH ₃ CH ₃ CH ₃	HO CH_3 H CH_3 H HO CH_3 H HO HO CH_3 H HO HO CH_3 H HO HO CH_3 H HO HO CH_3 H HO HO CH_3 H HO HO CH_3 H HO HO CH_3 H HO HO HO CH_3 H HO HO HO HO CH_3 H HO	HO HO CH_3 HO HO CH_3 HO HO HO CH_3 HO
Inhaler	Ointment / Cream	Ointment / Cream / Inhaler

Beclomethazone dipropionate:

- 5000 X as glucocoticoid as hydrocortisone.
- 500 x as betamethasone.
- Beclomethasone monopropionate g (17) is more active.

Diflucortolone valerate	Clobetasol propionate
HO CH_3 CH_3 CH_3 F F F F	HO CH ₃ CH
Ointment / Cream	Ointment / Cream

Anti-neoplastic agents

Neoplasm [Tumor] [Cancer]:

Relatively autonomous growth of tissues.

Types of Neoplasm:

There are various types of neoplasm BUT there is <u>no system for their nommenclature</u>: j Some neoplasms are <u>named according to their tissue or origin</u>. For example:

Tissue	Name of Tumor
<u>Epithelial</u>	Carcinoma
Connective tissue	Sarcoma
<u>Blood</u> [abnormal # in leukocytes]	Leukemia
Fibrous tissue	Fibroma

 ${f k}$ Others named after the individual who first describe the condition.

E.g. Hodgkin's disease.

Etiology of Cancer :

Many factors have been supposed to be the cause of cancer BUT none of them is the exact etiologic factor. E.g :

j <u>Genetic factors</u> : mutation & changed gene expression.

k <u>Viral factor</u> :

e.g. human T-lymphotropic virus type I [HTLV-I] " the cause of a form of leukemia.

- **1** <u>Physical factors</u> : as long term exposure to chemical and/or irradiation.
- m Hormonal factor.

Mechanism of Cancer formation :

By Mutation [Altered Gene Expression]

- <u>Proto-Oncogenes</u>: its function is to control cell growth.
- Oncogenes : normal proto-oncogene by mutation " incorrect proto-oncogene expression
- " Oncogene " synthesis of oncoprotein which is devoid of important regulatory elements
- " # DNA replication.
- <u>Anti-Oncogene or Supressor gene :</u> guard the cell \neq oncogene action.

<u>There are 2 fundamentally different genetic mechanisms exist :</u>

j Enhance or aberrant oncogene expression.

k \$ activity of anti-oncogenes " tumor suppressor genes".

Lines of treatment:

- <u>Surgical methods</u> : for large tumors "if it has not metastasized".
- , Radiotherapy :
 - Superior to surgery that it effectively destroys a tumor with minimal damage to surrounding normal tissues.
 - May be done after surgery to remove traces remain.

f <u>Chemotherapy</u> : [Antineoplastic, Anticancer, Antitumor or Cytotoxic Drugs]

Chemotherapeutic Agents

[Antineoplastic, Anticancer, Antitumor or Cytotoxic Drugs]

- Chemotherapy is not so much limited by metastasis.
- Ideal anti-cancer should theoretically eradicate cancer cells without harming normal cells.

• Anticancer act on cells in the process of division, so as tumor cells divide more rapidly than normal cells , they are most affected.

• That's why normal cells with high rate of division are also affected causing many <u>side effects</u> :for e.g.

• Hair follicles \rightarrow hair loss.

k Lymphatic system [which is responsible for the defense mechanism of the body \neq foreign matter] "decrease in immunity & # susceptibility to infection.

f Bone marrow \rightarrow aplastic anaemia, leukemia.

m Mucous membrane of the stomach " nausea & vomiting.

Cell Cycle Kinetics:

G_1	Pre-synthetic phase	
S	Synthetic phase	
G_2	Synthesis of cellular	
	components	
М	Mitotic phase	
Go	Resting phase	

<u>Classification of cytotoxic drugs according to chemical nature</u>, pharmacological action & <u>M.O.A</u>:

- Alkylating agents.
- , Anti-metabolites [Specific S].
- **f** DNA intercalating agents.
- " Antibiotics.
- ... Antimitotic agents [Specific M].
- † Hormones.
- p Miscellaneous cpds.

• <u>Alkylating agents</u>

- Act by alkylating DNA bases & nucleophilic attack on proteins.
- They includes:
- [a] Nitrogen mustards.
- [d] Nitrosoureas.

[b] Aziridines. [e] Triazines. [c] Alkyl sulphonates derivatives.

General formula:

History:

<u>World War I</u>: Usage of Sulphur mustard as <u>vesicant gas</u> [chemical warfare agent] \rightarrow it was found to have sever <u>lymphoid aplasia</u> as well as pulmonary irritation.

 $S_{CH_2 \cdot CH_2 \cdot CI}^{CH_2 \cdot CH_2 \cdot CI}$ (Dichoroethyl sulphide)

So, we started clinical trials to find related but less toxic & more soluble derivative This lead to development of nitrogen mustard from sulphur mustard **à** where, in addition to its vesicant effect on skin, they produce atrophy of lymphoid tissue & bone marrow.

<u>Uses :</u> limited to treatment of lymphomas especially Hodgkin's disease. <u>M.O.A.:</u>

Examples on M.O.A. of nitrogen mustard agents:

Alkylation of Guanine base of DNA:

Both aliphatic & aromatic nitrogen mustards react with 7-position of Guanine in each of double strands of DNA causing <u>cross-linking</u> [bifunctional alkylating agents], " interfere with separation of the strands " prevent mitosis.

SAR of nitrogen mustards :

- 1. Bis (2-chloroethyl) is essential for activity.
- 2. Halogen other than chloride à â activity.
- 3. Ethylene moiety between the nitrogen & chloride is Essential for activity à due to formation of aziridinium ion.

Methylene or trimethylene moiety \dot{a} abolish this activity.

Mechlorethamine	k <u>Melphalan</u>	l <u>Chlorambucil</u>	m Cyclophosphamide
[Mustine	[Alkeran [®]]	[Leukeran®]	[Endoxan [®]]
hydrochloride][HN2]			
H ₃ C-N CI <u>N-methyl-bis-[2-</u> <u>chloroethyl] amine</u>	 A-[bis (2-chloroethyl) amino] –L- phenylalanine Me too drug with phenylalanine substituting the methyl gp [e- donating] to N. Mechanism : It's selective for melanoma cells due to presence of phenyl alanine which is the precursor of melanine" so, uptaked by these cells" DNA alkylation. 	HOOC 2 4 HOOC 2 4 	$\frac{4}{5} \frac{3}{1} \frac{1}{CH_2-CH_2-CI} \frac{1}{5} \frac{1}{5} \frac{2}{5} \frac{1}{5} \frac{2}{5} \frac{1}{5} \frac{2}{5} \frac{1}{5} \frac{1}{5} \frac{2}{5} \frac{1}{5} \frac{1}{5} \frac{2}{5} \frac{1}{5} \frac{1}{5} \frac{1}{5} \frac{2}{5} \frac{1}{5} \frac{1}{5} \frac{1}{5} \frac{2}{5} \frac{1}{5} \frac$
Route : I.V. infusion " strong vesicant that can't be taken orally.	Orally as HCI	Orally	Orally & parentrally → adv. over other alkylating agent.
Uses : • In Hodgkin's disease & non- Hodgkin's lymphomas.	 j Skin cancer [melanoma cells]. Its designed ≠ melanoma tumor cells as phenylalanine derivative of nitrogen mustard. k Breast & ovarian cancer. 	 In chronic lymphocytic leukemia. Toxic effects: it's the <u>least toxic</u> <u>nitrogen mustard derivative, as it</u> <u>is the least active "act more</u> <u>slowly than others".</u> 	 Lymphomas, leukemias, sarcomas & carcinomas of breast or ovaries. Chlildhood malignancies.

Anti-Neoplastics

Synthesis of Mechlorethamine :

$$H_{3}C - NH_{2} + 2 CI - CH_{2} - CH_{2} - OH \xrightarrow{-2 HCI} H_{3}C - N \xrightarrow{CH_{2} - CH_{2} - OH} \underbrace{SOCI_{2}}_{CH_{2} - CH_{2} - OH} \xrightarrow{H_{3}C} N \xrightarrow{CH_{2} - CH_{2} - CI}_{CH_{2} - CH_{2} - CI}$$

Synthesis of Melphalan :

Synthesis of Chlormabucil :

Synthesis of Cyclophosphamide :

Biotransformation [Activation] of Cyclophosphamide:

Cyclophosphamide is inactive in vitro & must be converted to active form by metabolic process.

N.B. Acroline is nephrotoxic [cause cystitis], so, give N-acetylcystein or Mesna which are thiols to remove it as follow:

[b] Azir idines

- They are strained ring system [aziridine ring].
- Activity of aziridine gp # by protonation.

Thiotepa -N〔

N

N,N',N"- triethylene thiophosphoramide. OR Tris- [1-aziridinyl] phosphine sulphide.

Uses :

- Breast & ovarian carcinomas.
- **k** Malignant lymphoma.

N.B: thiotepa is most active at low pH.

[c] <u>Alkyl Sulphonatederivatives</u>

Busulphan [Myleran[®]]:

<u>M.O.A. :</u>

It undergo ionization to carbonium ion, then methyl sulphonate gp is displaced by $Nu \rightarrow Busulphan$ is a mono & bifunctional alkylating agent.

Synthesis of Busulfan:

Assay of Busulfan:

Uses:

Orally for chronic granulocytic leukemia.

[d] <u>Nitrosoureas</u>

General M.O.A :

Uses :

- 1. Due to its <u>high lipophilicity</u> \rightarrow it passes B.B.B. \rightarrow used in <u>brain carcinoma</u>.
- 2. 2ry therapy of Hodgkin's disease & other leukemias.

[e] <u>Triazines</u>

Dacarbazine [DTIC - Deticene]:

CONH ₂	
3 $^{N}_{H}$ $^{4}_{5}$ $^{CH_{3}}$	5- (3,3- <u>d</u> imethyl-1- <u>t</u> riazenyl) -1H- <u>i</u> midazole-4- <u>c</u> arboxamide
$2 \frac{1}{N}$ $N = N - N$	
H H	

M.O.A.:

Synthesis:

Uses : Parentrally [I.V]

- 1. For metastatic malignant melanoma.
- 2. In combination therapy for Hodgkin's disease.

, <u>Anti-metabolites</u>

Definition:

- They are substances of <u>structural similarity to normal metabolites</u> → interfere with the utilization & biosynthesis of normal metabolites [as folate, purine & pyrimidine pathways that are required for DNA synthesis].
- Its selectivity \neq cancer cells is <u>due to differences in cell growth fractions</u>.

[a] Dihydrofolate reductase inhibitors

Role of folic acid:

Anti-Neoplastics

- Interfere with purine biosynthesis & purine interconversion.
- Used orally for acute leukemia.
- Both are inactive until metabolized to respective monophosphate ribonucleotides by action of <u>Hypoxanthine-Guanine Phosphoribosyl transferase</u> [HGPRTase]

```
M.O.A. of 6-MP:
```

1) Activation by HGPRTase to form active <u>6-Thioinosinate</u> [6-TI]

2) 6-TI is a potent inhibitor of conversion of 5-phosphoribosyl phosphate to 5-phosphoribosylamine [Rate controlling step in purine synthesis]
3) 6-TI à also inhibit conversion of of Inosine Monophosphate [IMP] to Adenine Monophosphate [AMP] & Xanthine Monophosphate [XMP] & limits availability of XMP to form Guanine Monophosphate [GMP] à interfere with supply of purine precursors for nucleic acid synthesis.

Catabolism of 6-MP:

- 1) By oxidation of 6-TI à <u>6-Thiouric acid</u> [Gout & Nephrotoxicity].
- 2) So, Allopurinol is co-given à it's Xanthine Oxidase Inhibitor.
- If give Allopurinol à inhibit metabolism of 6-MP à á toxicity, so, reduce dose of 6-MP by 25-30 %.

6-Thioguanine:

- When allopurinol is co-given with it [in case if cancerous patient is also with gout] à <u>No need for dose reduction</u>.
- <u>With the same M.O.A. of 6-MP</u>, differ only in: It's converted further into di- & triphosphate à incorporated in RNA structure & its deoxy form incorporated in DNA structure instead of normal Guanine à <u>altered polynucleotide</u>.

Azathioprine	
$H_{3}C \longrightarrow NO_{2}$	 Ø It's <u>designed to protect 6-MP from</u> <u>catabolic reaction</u>. Ø Not have better anti-cancer activity but <u>used as immuno-suppressive</u> in organ transplantation.

[c] Pyrimidine Antagonists

<u>M.O.A.:</u>

• It's <u>anti-metabolite for thymine & uracil</u> [Pyrimidine bases of DNA]: <u>C-F bond is extremely stable & prevent addition of methyl gp in 5-position</u> " no formation of thymidine.

, Flurouridine monophosphate [FUMP] converted to triphosphate [FUTP] form which is incorporated in RNA causing error in base pairing during RNA transcription.

<u>Uses</u> : Parentrally [I.V] (although some 5-FU are absorbed orally)

for carcinoma of : breast, stomach, pancreas, colon & rectum in patients who can't be cured by surgery or other means.

Anti-Neoplastics

f<u>Antibiotics</u>

	[a] <u>Anthracyclines</u>	[b] <u>Mitomycin</u>
Structure	$R = CH_2OH \longrightarrow Daunorubicin$ $R = CH_3 \longrightarrow Daunorubicin$	$H_2N \xrightarrow{O} O \xrightarrow{NH_2} O \xrightarrow{H_2N} O \xrightarrow{VH_3} O $
	isolated from Streptomyces peucetius.	
M.O.A.	 <u>DNA-RNA binding :</u> they intercalate between base pairs on the double helix of DNA " partial unwinding of the helix " disrupt DNA polymerases & transcription. <u>Free Racidal Generation</u> : F.R. formation which is highly reactive compounds with unpaired electrons " this occurs during the metabolism of anthracyclines. <u>Membrane interaction :</u> anthracyclines bind to cell membrane " change in membrane glycoproteins, transmembrane flux of ions & membrane morphology. <u>Metal ion chelation :</u> chelate divalent cations as Ca⁺² & Fe⁺² ions by quinine & phenolic functions. 	$H_{2}N \rightarrow H_{2}$ $H_{2}N \rightarrow H_{2}$ $H_{3}C \rightarrow OCH_{3}$ $H_{2}N \rightarrow OH$ $H_{2}N \rightarrow OH$ $H_{2}N \rightarrow OH$ $H_{2}N \rightarrow OH$ $H_{3}C \rightarrow OH$ $H_{2}N \rightarrow OH$
	<u>Uses :</u> • Doxorubicin[Adriamycin [®]]" <u>PARENTRALLY</u> used in : 1. Solid tumors as carcinoma of breast, ovary, lung & thyroid. 2. Soft tissues sarcoma.	$H_{2N} \rightarrow H_{3C} \rightarrow 0$
	• Daunorubicin[Daunomycin [®]]" <u>PARENTRALLY</u> in Acute lymphocytic & granulocytic leukemia.	

"<u>Aniti-MitoticAgents</u>

	<u>Vinca alkaloids</u>	k Etoposide	l <u>Taxanes</u> Paclitaxel [Taxol [®]]
	• They are <u>Vincristine</u> & <u>Vinblastin</u> . • Family of important anti-tumor agents <u>from plants</u> . • With complex structure containing <u>Indole Ring</u> . • With complex structure containing <u>Indole Ring</u> . • With complex structure containing <u>Indole Ring</u> . • Vincristine : $R = CHO$ Vincristine : $R = CHO$ Vinblastin : $R = CH_3$ H_3CO	Etoposide is semi-synthetic derivatives of podophyllotoxin which is a cytotoxic drug isolated from the root of May apple plant.	 One of the most important new classes of anti-tumor drugs. Paclitaxel has been marketed in 1994 as <u>Taxol[®]</u>. Isolated from the bark of yew tree & was produced semi- synthetically.
<u>M.O.A</u>	Bind to critical <u>microtubular proteins</u> within the cell which are essential contractile proteins of the mitotic spindle of dividing cells " <u>mitotic arrest</u> .	Cause metaphase arrest of the cell cylcle.	Act ≠ spindle assembly of dividing cells <u>but in</u> <u>different fashion from</u> that of Vinca alkaloids.
<u>Uses</u>	<u>Vinblastin SO4</u> [Velban [®]] [I.V.]: Used in testicular carcinoma. <u>Vincristine SO4</u> [Oncovin [®]] [I.V.]: • Acute lymphocytic leukemia. , Lymphomas, breast cancer, sarcomas & various childhood neoplasms.	Used in: [Orally & I.V] j Combination therapy ofTesticular carcinoma & aggressive forms of non- Hodgkin's lymphomas. k The most active single agent used in small-cell lung cancer.	 <u>Used I.V. in :</u> Ovarian & breast cancer. Lung cancer & other cancers.

<u>nHormones</u>			
j Anti-estrogens	k Estrogens		l Progestins
<u>Tamoxifen</u>	Diethylstilbsterol [DES]	Estramustin Na ₂ PO ₄	Megsterol acetate
H ₃ C ^N CH ₃ O	HO	OPO ₃ Na ₂	OCOCH3 OCOCH3 CH3
Used in ttt of <u>advanced</u>	Synthetic estrogen for	 It's composed a of 	<u>Medroxyprogesterone</u>
postmenopausal women.	prostate cancer.	<u>estradiol + phosphate</u>	<u>acetate</u>
<u>N.B.</u> # estrogen " breast cancer.	• <u>S.E. feminization males.</u> So, <u>Fosfosterol</u> is used w' target it to prostate tissue, as it's <u>rich</u> in phosphatase enz. $(\downarrow \downarrow $	 <u>estradior + prosprate</u> <u>ester.</u> So, it's designed to targe nitrogen mustard to cells with estrogen receptors BUT it was found to act by Anit-Androgenic effect NOT by Alkylation. Used for <u>Prostatic</u> <u>Carcinoma</u> 	$ \begin{array}{c} & \underset{CH_3}{\overset{O}{\leftarrow}} \\ & \underset{CH_3}{\overset{O}{\leftarrow}} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $

o <u>Miscellaneous compounds</u>

	Heavy Metal Complexes		Hydroxyuroa	
	<u>Cisplatinum</u> [Cisplatin [®]]	Carboplatin [Paraplatin®]	[Hydrea [®]]	
Structure	H ₃ N Pt CI CI CI CI CI CI CI CI CI CI	<u>Cis-diamine(1,1-cyclobutane-dicarboxalato) platinum</u>		
	The first heavy metal to be introduced as anti-cancer.	Second generation cisplatin analogue with less nephorotoxic, neurotoxic & emesis action than parent compound.	H ₂ N ^C NH-OH	
M.O.A	Potent inhibitor of DNA polymerase. <u>Its activity & toxicity resemble</u> <u>alkylating agents</u> " the 2 chlorine atoms are active & displaced by nitrogen or oxygen atoms of purines [Cross-linking].		Inhibit <u>ribonucleotide reductase</u> [which converts ribonucleotide to deoxyribonucleotide] " \$ deoxyribonucleotide " inhibit DNA synthesis. * It may interfere with function of the enzyme by chelating with its Fe ⁺⁺ cofactor.	
Uses	 Parentrally in combination for : Testicular & ovarian carcinoma. Head & neck cancer. [highly active & curative]. 	• Ovarian cancer.	<u>Orally</u> for : • Melanoma. • Chronic myelocytic leukemia. • Metastatic ovarian carcinoma.	

p<u>New anti-neoblastic agents</u> <u>Signal Transduction Inhibitors</u>

• Used in treatment of other cancers where EGFR over expression is involved.